

Barium Metaplumbate Thin Film Electrodes for Ferroelectric Devices

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Barium metaplumbate thin films were deposited *in situ* by pulsed laser deposition on Si/SiO₂ and Si/SiO₂/Ti/Pt substrates between 400°C and 700°C. Films prepared at low temperature ($\leq 500^\circ\text{C}$) over platinum were randomly oriented. For depositions made at 2×10^{-2} mbar of oxygen, the films were oriented (110) for all temperatures, while at 0.1 mbar the films were oriented (110) at 500°C, changing to a mixed (222)/(200) orientation above 550°C. The conductivity of BaPbO₃ reached values of $5.6 \times 10^{-5} \Omega \text{ cm}$. The films deposited directly over silica were polycrystalline for temperatures above 500°C, having a strong (110) orientation only at 700°C. The orientation of BaPbO₃ deposited either on silica or platinum, was reflected on the PZT films deposited at room temperature over BaPbO₃ and crystallized by different thermal treatments. PZT capacitors made over BaPbO₃ presented high values of remnant polarization (up to $44 \mu\text{C}/\text{cm}^2$).

Keywords: Laser ablation; thin films; conductive oxides; electrodes; capacitors; ferroelectric properties

1. INTRODUCTION

Barium metaplumbate BaPbO₃ (abbreviated as BPO) is a useful material in many applications in microelectronics due to its metallic conduction behavior, with a low resistivity and a small positive temperature coefficient. Ikushima and Hayakawa [1] explained these phenomena as being derived from oxygen vacancies while Shannon and Bierstedt [2] considered the filled d bands of Pb⁴⁺ to be the real reason. Electrical and stability characteristics of BPO were studied in bulk ceramics by Hsieh and Fu [3].

A few results on BPO thin films deposited by RF magnetron sputtering have been recently reported by Luo and Wu [4], but no data has yet been published on BPO thin films made by pulsed laser deposition (PLD). The study of BPO thin films is important because of their possible application as electrodes in ferroelectric devices. The orthorhombic perovskite structure of BPO makes it suitable for being used as templates for growing thin films of ferroelectric materials with similar structure, like lead zirconate titanate (PZT). PZT thin films are used in capacitors for ferroelectric random access memories due to their nonvolatility and fast switching speed. However, the application of these ferroelectric capacitors is limited by the fatigue phenomenon, defined as the gradual decrease of the switching polarization with the number of switching cycles when platinum electrodes are used [5]. The mechanisms responsible for the fatigue phenomenon are still not fully understood, but electromigration of charged defects and/or charge injection into the ferroelectric film are probably involved [6].

Many efforts have been focused on retaining long term reliability of ferroelectric capacitors by finding optimal oxide electrode materials; RuO₂, La_{1-x}Sr_xCoO₃, LaNiO₃ and other oxides associated with platinum have been proposed [7–9]. Oxide/PZT/oxide heterostructures have excellent resistance to polarization fatigue, but they usually have large leakage currents and are more susceptible to dielectric breakdown. RuO₂ electrodes, for example, exhibit large and variable leakage currents (10^{-4} - 10^{-3} A/cm² at 1V) [10]. Despite the good fatigue performance of PZT capacitors using La_{1-x}Sr_xCoO₃ top and bottom electrodes, their remnant polarization is lower than that of Pt/PZT/Pt capacitors and they require high temperatures for crystallization heat treatment [11].

In the present work the pulsed laser deposition of BPO *in situ* on Si/SiO₂ and Si/SiO₂/Ti/Pt substrates was investigated. The effects of different deposition conditions and heat treatments on the properties of the PZT films were also studied.

2. EXPERIMENTAL DETAILS

For obtaining SiO₂ on Si, the Si substrates were heated at 950°C for 72 h in wet O₂ flow. Ti and Pt films were deposited using the RF sputtering technique in Ar with 2 in. diameter metal targets; the sputtering chamber had a base pressure of 10⁻⁶ mbar. Ti films 20 nm thick were deposited at room temperature using an RF power of 150 W and a deposition pressure of 10⁻² mbar. The Pt films 200 nm thick were sputtered at 500°C with the same RF power and the same deposition pressure.

The targets for laser ablation, BPO and PZT (Zr/Ti ratio = 52/48), were 1.5 cm diameter dense ceramic disks. The ceramic powders were prepared by the conventional mixed oxide method. For the BPO targets, stoichiometric proportions of barium carbonate and lead oxide were mixed and calcined at 700°C for 2 hour, to obtain the pure BPO phase. For the PZT (52/48) targets, stoichiometric proportions of lead carbonate, zirconium and titanium oxide were mixed and calcined at 900°C for 1 hour, to form the pure PZT phase. Fine BPO and PZT powders ($<5 \mu\text{m}$) were obtained after milling for several hours in a planetary mill. Circular pellets were prepared by uniaxially pressing at 200 MPa. In order to obtain dense BPO and PZT ceramics, the BPO samples were sintered in air between 800°C and 1000°C for 2 hour in the presence of BPO packing powder; the PZT samples were sintered at 1250°C for 2 hour, with a O_2 flux and in the presence of lead zirconate packing powder.

Before depositing each film the chamber was evacuated until at least 10^{-7} mbar. For all the depositions, a KrF laser with energy of 200 mJ for a 10 Hz repetition rate and a fluence of 1.4 J/cm^2 was used. The BPO films were deposited in an O_2 atmosphere at two different pressures: 1×10^{-1} mbar and 2×10^{-2} mbar; the distance between target and substrate was 4 cm and the deposition time was 10 minutes. Under these conditions, the deposition rate of BPO was 45nm/min.

Two series of BPO films were deposited *in situ* with different deposition conditions. The first series was made on Si/SiO₂/Ti/Pt and the second one was deposited directly over Si/SiO₂. The temperatures for the depositions made over platinum ranged between 400°C and 700°C with a step of 50°C while the temperatures for the depositions over Si/SiO₂ ranged between 450°C and 700°C with a step of 50°C. Without breaking the vacuum in the PLD chamber after deposition of BPO, the target was replaced with a PZT target (Zr/Ti ratio = 52/48) and PZT films were deposited on top of BPO at room temperature with a deposition pressure of 2×10^{-2} mbar in O_2 . The energy of the laser was 300 mJ for a 10 Hz repetition rate and the deposition time was 30 minutes with a target-substrate distance of 4 cm.

For crystallizing the PZT films, two methods were used: furnace annealing at 675°C for 15 minutes with heating and cooling rates of 50°C/min and rapid thermal annealing (RTA) at 650°C with a heating rate of 20°C/sec. Aluminum top electrodes were deposited by thermal evaporation at a pressure of 10^{-6} mbar, using a shadow mask.

The thickness of the films was measured with a contact profilometer (Dektak II A) and they were structurally analyzed using optical and scanning electron microscopy (JEOL JSM-6301F) and X-ray diffraction (Siemens

D5000). A system composed of an amplifier, a function generator (HP 8116A), an oscilloscope (LeCroy 9310 M) and a computer running a Lab-View program was used for recording the hysteresis loops. Electrical resistivity was measured on the BPO films deposited over both Pt and SiO₂ using the four-probe technique.

3. RESULTS AND DISCUSSION

3.1. BPO Films

For the deposition of BPO films with stable properties it is important to use fully reacted targets sintered at high temperature. Figure 1 shows X-ray diffraction patterns for BPO targets sintered at 800°C and 1000°C. One can see from the diffractograms that the target sintered at low temperature has residual barium and lead oxide peaks. Films made from this target were highly hygroscopic and in less than one minute the presence of a reacted layer could be observed on the surface of the films as shown on Fig. 2.

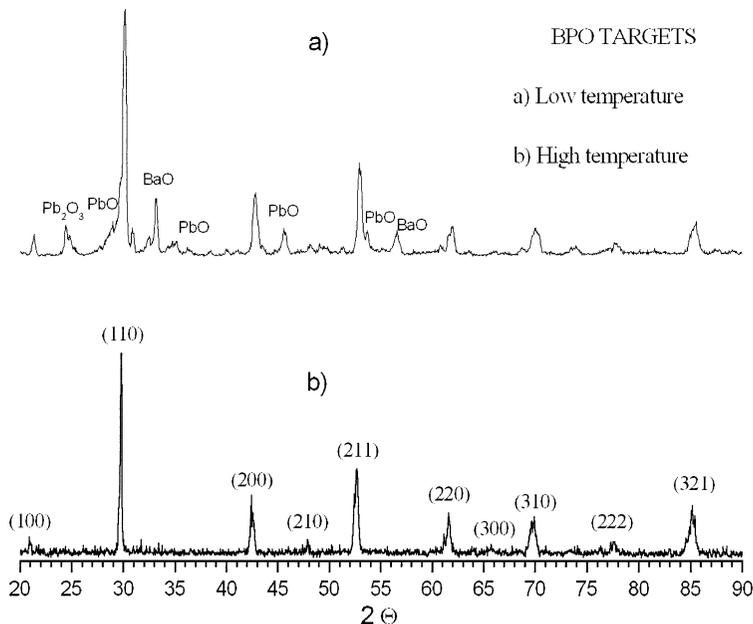


Figure 1. X-ray diffraction pattern of the BPO targets sintered below (a) and at (b) 1000°C.

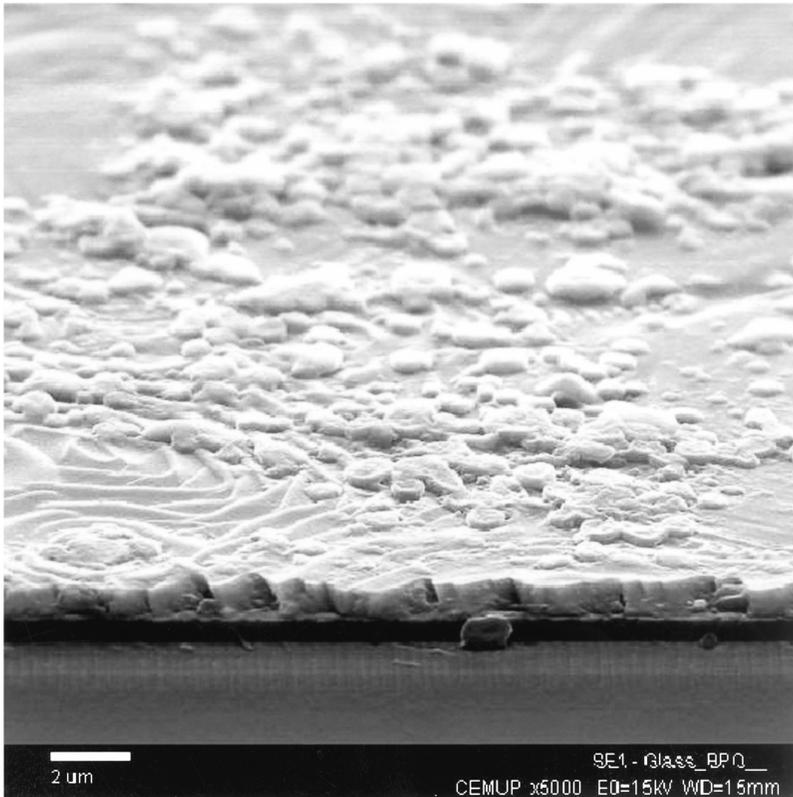


Figure 2. Reacted layer on the surface of BPO film due to atmospheric moisture.

The X-ray diffractogram of the BPO target sintered at 1000°C shows only the peaks from BPO and the films made from this target were immune to the atmospheric moisture. Figure 3 shows an optical micrograph of the surface of a film made at 600°C over platinum; the BPO films are black, smooth, uniform and no reactions could be detected, even after long term air exposure.

The BPO films made over Pt have shown a crystalline orientation dependent on the deposition temperature. In Fig. 4 the X-ray diffraction patterns for three significant temperatures are plotted. At 400°C the BPO is not crystallized yet; with the increase of the temperature, the BPO peaks start to appear and the films are randomly oriented. Increasing even more the deposition temperature, the preferred orientation starts to be (110) at 500°C

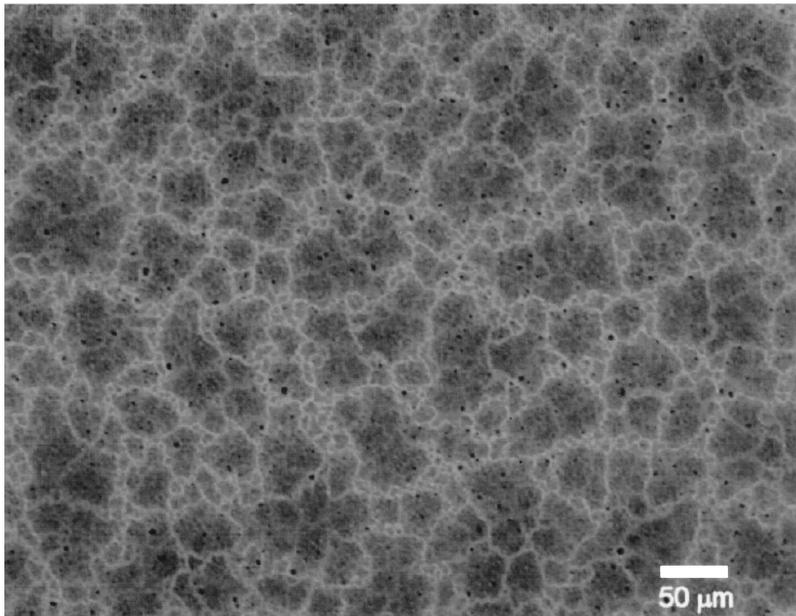


Figure 3. Surface of a BPO film made from target sintered at high temperature.

and the (222) peak appears. For higher temperatures the (222) peak and the $\langle 100 \rangle$ peaks become important while the $\langle 110 \rangle$ orientations are weaker. The (222) orientation is promoted by the platinum layer which is (111) oriented.

The effect of Pt is evident in the analysis of the BPO films made directly over SiO_2 at the same deposition temperatures (Fig. 5). In the X-ray diffractograms of the films made without Pt one can see that the (222) orientation of BPO is not present for any deposition temperature. In this case the temperature for crystallizing the BPO is increased to values over 450°C due to the absence of crystalline orientation in the underlying silica. For temperatures higher than the crystallization temperature, the BPO films are oriented (110) and (100) for 600°C , while at 700°C the film is strongly (110) oriented. In the absence of Pt all the temperatures needed for obtaining the different orientations in BPO films were increased by at least 50°C .

Electrical resistivity was measured on the BPO films deposited over both Pt and SiO_2 . The BPO films deposited over platinum exhibited resistivities ranging between $1.1 \times 10^{-4} \Omega \text{ cm}$ and $5.6 \times 10^{-5} \Omega \text{ cm}$. For the BPO films deposited over silica the resistivities were between $2.5 \times 10^{-1} \Omega \text{ cm}$ and $3.6 \times 10^{-1} \Omega \text{ cm}$. For assessing the performance of BPO as bottom electrode

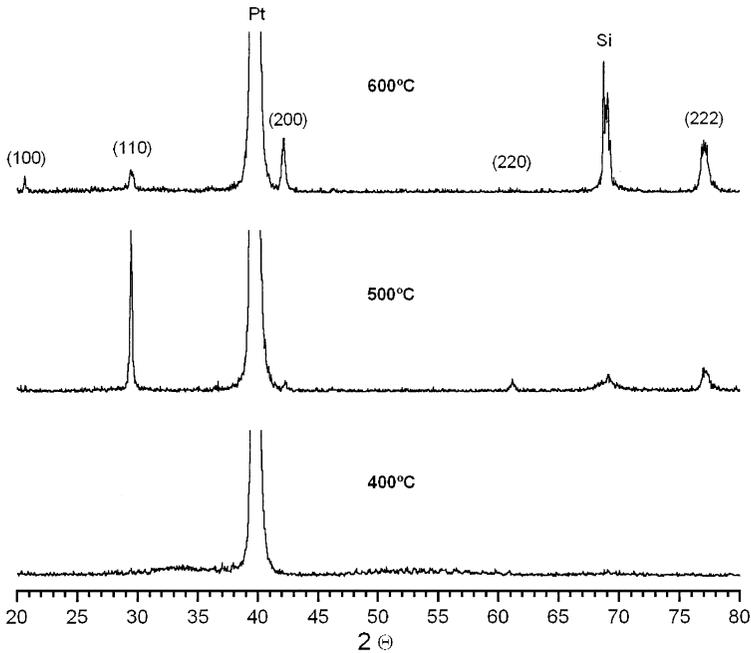


Figure 4. X-ray diffraction pattern of BPO films made over Pt.

in ferroelectric capacitors, PZT films were deposited by laser ablation on top of the BPO films.

3.2. PZT Films

The orientation of the BPO and PZT films after furnace crystallization of the PZT at 675°C was analyzed by X-ray diffraction (Figs. 6 and 7). In both graphs the families of planes are shown in brackets, and one can see that the peaks for BPO and PZT appear in pairs and their intensities are similar. Another important feature of the films made by laser ablation is the absence of peaks from pyrochlore, in contrast with the results reported by Luo and Wu on films made by sputtering in the same temperature range [4]. Figure 6 shows the X-ray diffraction pattern for the PZT crystallized over BPO made at 550°C and 10^{-1} mbar O_2 .

One can see clearly from the graph that the BPO is strongly oriented in the (222) direction and the PZT follows the BPO orientation, indicating epitaxial growth of the PZT. With the decrease of the deposition

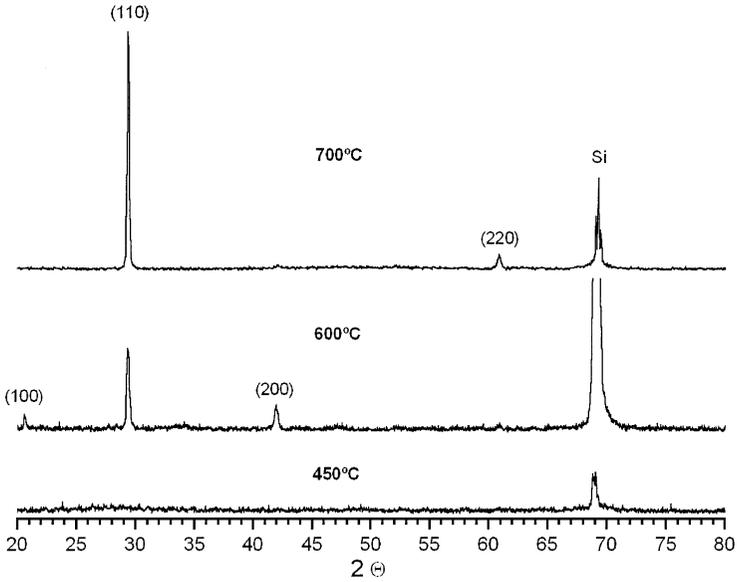


Figure 5. X-ray diffraction pattern of BPO films made over SiO₂.

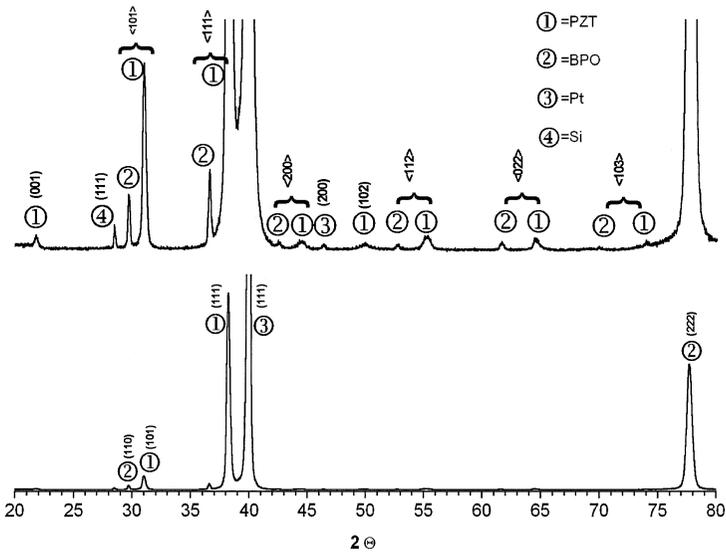


Figure 6. X-ray diffraction pattern for the PZT crystallized over BPO made at 550°C and 10⁻¹ mbar O₂. The upper graph is the same as the lower one but with the vertical scale increased.

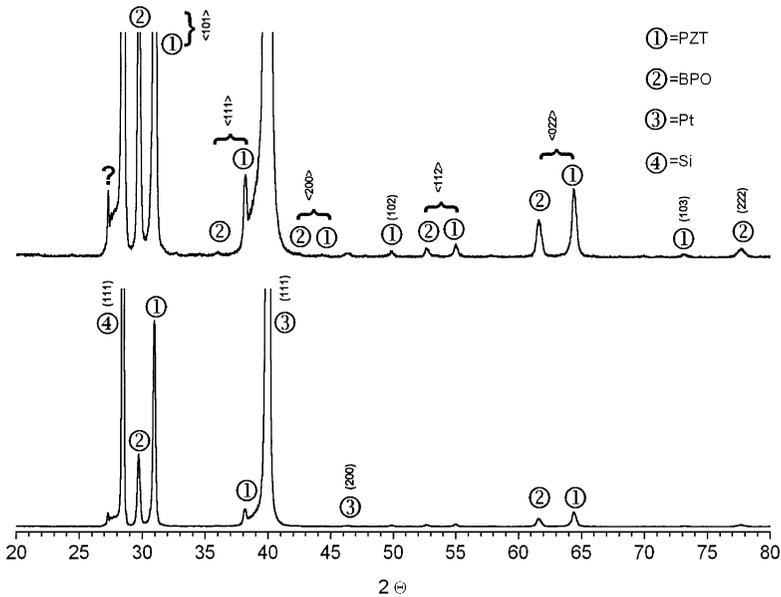


Figure 7. X-ray diffraction pattern for the PZT crystallized over BPO made at 550°C and 2×10^{-2} mbar O_2 . The upper graph is the same as the lower one but with the vertical scale increased.

pressure of BPO the orientation changes to (110), as shown in Fig. 7, and again the PZT is following the same trend, in this case with a (101) orientation.

A PZT film deposited over SiO_2/BPO and crystallized by RTA at 650°C is shown in Fig. 8. One can see that the BPO is adherent to silica and the surface of the PZT is smooth with grains of around 1 μm . The boundary between BPO and PZT films is clearly visible on the fracture.

Figure 9 shows the X-ray diffractograms for PZT films deposited over SiO_2/BPO made *in situ* at 550°C and 650°C and crystallized by RTA at 650°C. For BPO films made at 550°C both BPO and PZT are polycrystalline, without a dominant orientation, while for BPO films made at 650°C both BPO and PZT have orientations belonging to the {110} family. The smaller peaks in the lower graph are all from the other orientations of BPO and PZT.

Ferroelectric properties were investigated for samples where the PZT was deposited over different types of electrodes: Pt/BPO, BPO and Pt. In Fig. 10 are shown the hysteresis loops for BPO deposited over Pt and PZT crystallized in the furnace at 675°C for 15 min, BPO

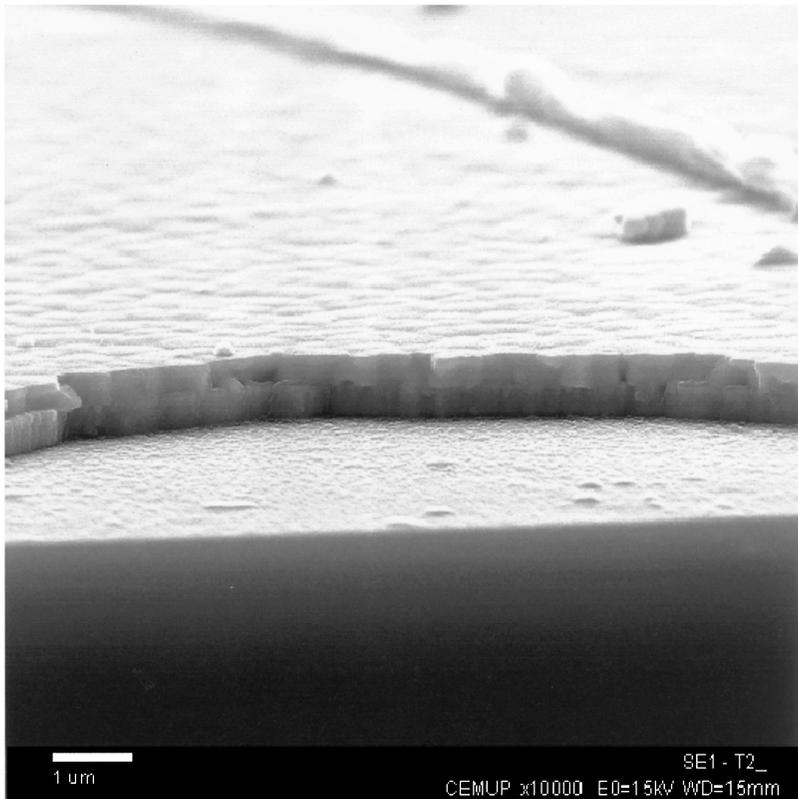


Figure 8. Cross-section of a PZT film deposited over SiO₂/BPO and crystallized by RTA at 650°C.

deposited directly over silica and PZT crystallized by RTA at 650°C and for PZT deposited over Pt and crystallized in the furnace at 675°C for 15 min.

The results for the PZT films deposited on Pt and Pt/BPO are better than the ones for films deposited over SiO₂/BPO due to the PZT orientation which is (111) when the Pt is present and (101) in the absence of Pt. The remnant polarization for films deposited over Pt/BPO is very high (44 μC/cm²) but the coercive field is also high (145 kV/cm). The coercive field became comparable with the Pt/PZT films for samples made directly over SiO₂/BPO.

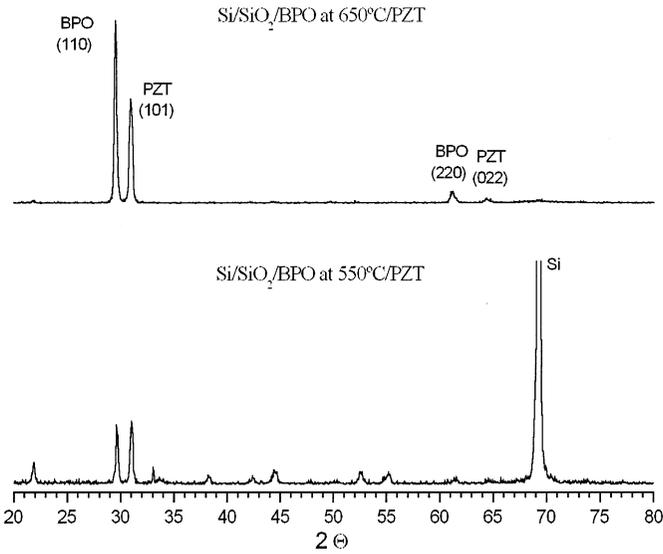


Figure 9. X-ray patterns for PZT films deposited over SiO₂/BPO made *in situ* at 550°C and 650°C and crystallized by RTA at 650°C.

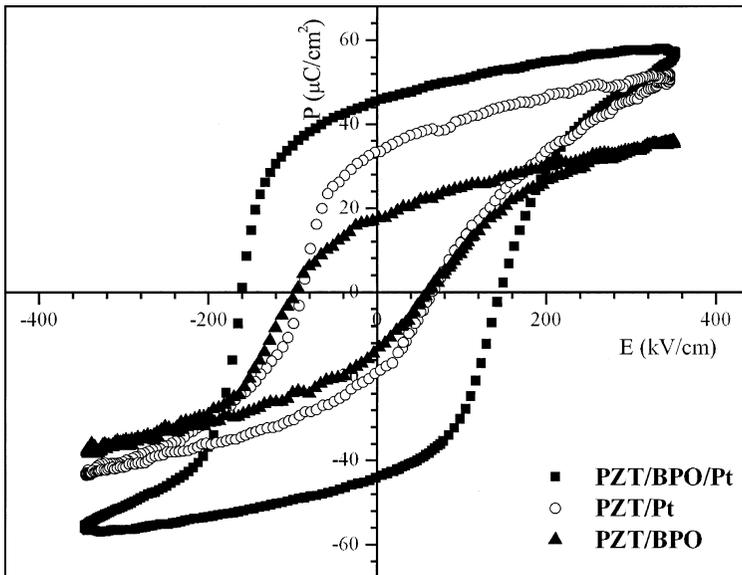


Figure 10. Hysteresis loops for PZT films deposited over Pt/BPO, BPO and Pt electrodes.

4. CONCLUSIONS

In summary, barium metaplumbate thin films were deposited by laser ablation with a high deposition rate (45 nm/min). BPO films made from targets sintered at low temperature are hygroscopic, but by using targets sintered at 1000°C stable, non-hygroscopic and conductive films could be obtained. The films deposited directly over silica were polycrystalline for temperatures above 500°C, having a strong (110) orientation only at 700°C. When deposited over platinum, the BPO made at 2×10^{-2} mbar of oxygen was oriented only in the (110) direction. For depositions made at 0.1 mbar of oxygen the films were oriented (110) at 500°C, changing to a mixed (222)/(200) orientation above 550°C. The orientation of the BPO films was reflected on the PZT films deposited at room temperature over BPO and crystallized after the deposition. Two different thermal treatments were applied for crystallization of the PZT: furnace annealing and rapid thermal annealing, with the best results in terms of ferroelectric characteristics (higher remnant polarization, lower coercive field) being obtained with furnace annealing. Ferroelectric capacitors made with PZT crystallized by furnace annealing over BPO deposited at 600°C and 0.1 mbar presented high values of remnant polarization ($44 \mu\text{C}/\text{cm}^2$).

REFERENCES

- [1] H. Ikushima and S. Hayakawa, *Solid-State Electron* **9**, 921 (1966).
- [2] R. D. Shanon and P. E. Bierstedt, *J. Am. Ceram. Soc.* **53**, 635 (1970).
- [3] Y. Hsieh and S. Fu, *Ceramics International* **18**, 289 (1992).
- [4] Y. R. Luo and J. M. Wu, *Appl. Phys. Lett.* **79**, 3669 (2001).
- [5] H. M. Duiker, P. D. Beale, J. F. Scott, C. A. Paz de Araujo, B. M. Melnick, J. D. Cuchlaro, and L. D. McMillan, *J. Appl. Phys.* **68**, 5783 (1990).
- [6] S. B. Desu, *Phys. Stat. Sol. A* **151**, 467 (1995).
- [7] G. Chao and J. Wu, *Jpn. J. Appl. Phys.* **40**, 2417 (2001).
- [8] M. S. Chen, T. B. Wu, and J. M. Wu, *Appl. Phys. Lett.* **68**, 1430 (1996).
- [9] M. S. Chen, J. M. Wu, and T. B. Wu, *Jpn. J. Appl. Phys.* **34**, 4870 (1995).
- [10] T. Morimoto, O. Hidaka, K. Yamakawa, O. Arisumi, H. Kanaya, T. Iwamoto, Y. Kumura, I. Kunishima, and S. Tanaka, *Jpn. J. Appl. Phys.* **39**, 2110 (2000).
- [11] R. Ramesh, T. Sands, V. G. Keramidas, and D. K. Fork, *Mat. Res. Soc. Symp. Proc.* **310**, 195 (1993).